Static Electricity

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TECHNICAL INFORMATION

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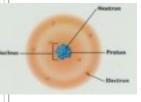


What is static electricity?

"Static Electricity is the imbalance of positive and negative charges"

Static electricity refers to the accumulation of excess electric charge in a region with poor electrical conductivity, such that the charge accumulation persists.

"To understand static electricity, we have to learn a bit about the nature of matter"



A t o m Everything around us is made of atoms. There are 115 identified by the

different kind of atoms identified by the scientists so far. Everything you see is



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made of different combination of these atoms.

In the middle of each atom is a nucleus. The nucleus contains two kind of tiny particles, called protons and neutrons. Orbiting around the nucleus are even smaller particles called electrons.

The protons and neutrons in the nucleus are held together very tightly. Normally the nucleus does not change. But some of the outer electrons are held very loosely. They can move from one atom to another.

An atom that looses electrons has more positive charges (protons) than negative charges (electrons). It is positively charged. An atom that gains electrons has more negative than positive particles. It has a negative charge. A charged atom is called an "ion".

Cause of static electricity Electrons can be exchanged between materials on contact. Materials with

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You walk across the rug, reach for the doorknob ZAP!!! You get a static shock.

weakly bound electrons tend to lose them, while materials with sparsely filled outer shells tend to gain them. This is known as the **triboelectric effect** (Figure 1) and results in one material becoming positively charged and the other negatively charged.

The polarity and strength of the charge on a materials once they are separated depends on their relative positions in the triboelectric series. This is the main cause of static electricity as observed in every day life.



Triboelectric series

Most positively charged (+)

+



Most negatively charged (-)

Figure 1

Static electricity in our everyday life



"When two non-conducting materials are rubbed or be in contact with each other, a charge imbalance will occur after the two materials are separated"

Triboelectric effect

A material towards the bottom of the series, when touched to a material near the top of the series, will attain a more negative charge, and

vice versa. The further away two materials are from each other on the series, the greater the charge transferred.

"tribos" means

"rubbing" in the Greek

charge. Things with the same charge repel each other. So the hairs try to get as far from each other as possible. The farthest they can get is by standing up and away from the others. And that is

bad hair day!

Another example is as you walk

across a carpet , electrons move from the

rug to you. Now you have extra electrons



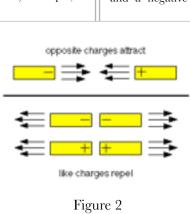
Opposites attract

Two materials with opposite,

or different charges (a positive and a negative) will attract, or pull towards each other. Things with the same charge (two positives or two negatives) will repel,

or push away from each other. (Figure 2)

What does all this have to do with our everyday life? When you take off your wool hat, it rubs against your hair. Electrons move from your hair to the hat. A static charge builds up and now each of the hairs has the same positive



and a negative static charge. Touch a door knob and ZAP! The door k n o b i s а conductor. The electrons jump from you to the knob, and you feel

the static shock.



how static electricity causes a

How does **CeraSOL** help reduce static electricity?

Presently, most hair brush, hair straightener and steam iron are made in or coated with plastic materials such as Teflon. As you can check from the Triboelectric series, Teflon and various plastic materials are in the bottom of the series, which means they are apt to "capture" electrons from your hair and clothes when they are in contact. As a result, static electricity is built up. Your hair will stand on end, and your clothes have problems of cling and sparking during ironing.

CeraSOL ceramic coating reduces the static electricity effect by effective emission of negative ion. The continuous supply of negative ion neutralize the positively charged hairs created through brushing or rubbing.

CeraSOL also reduces noises of sparking during ironing. Spark is the result of an electrical discharge of the build-up static electricity created when the Tefloncoated soleplate rubs against the clothes.

Furthermore, CeraSOL ceramic coating is ranked in the upper triboelectric series, which means it will not attain as much electrons from hairs or clothes as Teflon or any other resin-based coating. Thus, less static electricity is built up.

Tourmaline



Tourmaline crystal

"Tourmaline is a crystal silicate mineral that works perfectly with CeraSOL inorganic coating to generate negative ions."

The Magic of **CeraSOL**

"CeraSOL ceramic coating system shows much better negative ion emission efficiency due to its micro-pores structure." said Mr. S.M. Kim, top ceramic expert from Korea.

CeraSOL ceramic structure contains numerous micro pores (from 2.5 to 50 nanometers).

Kim explained "From the principle of negative ion emission, excessive electrons from tourmaline must interact with the water molecules from the air to generate negative ion. The most important variables for this movement is the mobility of electrons and water molecules."



"CeraSOL unique structure provides effective diffusion media for water molecules to come into contact with tourmaline to generate negative ion on continuous basis. Traditional resin-based coating system has very dense structure that

virtually blocks tourmaline from reacting with water molecules" added Kim

The manual states	Hair Brush	Iron
Iair Straightener		
Iair Straightener TEST ITEMS		
TEST ITEMS	AMOUNT	A UNIT